1. Using the Lewis structuress and polar bonds from our previous class exercise, show the correct 3-D structure of the following molecules using VSEPR Theory. Also indicate ALL polar bonds in the molecule and the direction of the molecular dipole moment for each molecule (if any).

a) NH₂Cl

b) ClF₃

c) XeF₄

d) CH₂O

e) CH₃

f) NH₄⁺