1. Predict the major product (or products) for each of the following reactions? In each part give the mechanism  $(S_n 1, S_n 2, E1, or E2)$  by which each product is formed.



2. Many biochemical processes occur by reaction pathways analogous to those carried out in the laboratory. Some toxic compounds are very good  $S_N^2$  substrates because they transfer alkyl groups to enzymes, thus deactivating them.

Enzyme–NH<sub>2</sub> + R–X – Enzyme–NHR + HX

Mustard gas in an infamous toxic thioalkylating agent,  $ClCH_2CH_2SCH_2CH_2Cl$ , that was responsible for 500,000 casualties in World War I. The overall reaction for mustard gas with an enzyme is as follows, and requires **two**  $S_N^2$  reactions and proceeds through an intermediate cation.

$$CICH_{2}CH_{2}CH_{2}CH_{2}CH_{2}CH_{2}CH_{2}CH_{2}CH_{2}N^{+}H_{2}-Enzyme$$

Use curved arrows to clearly show **both**  $S_N 2$  reactions and identify the intermediate.